



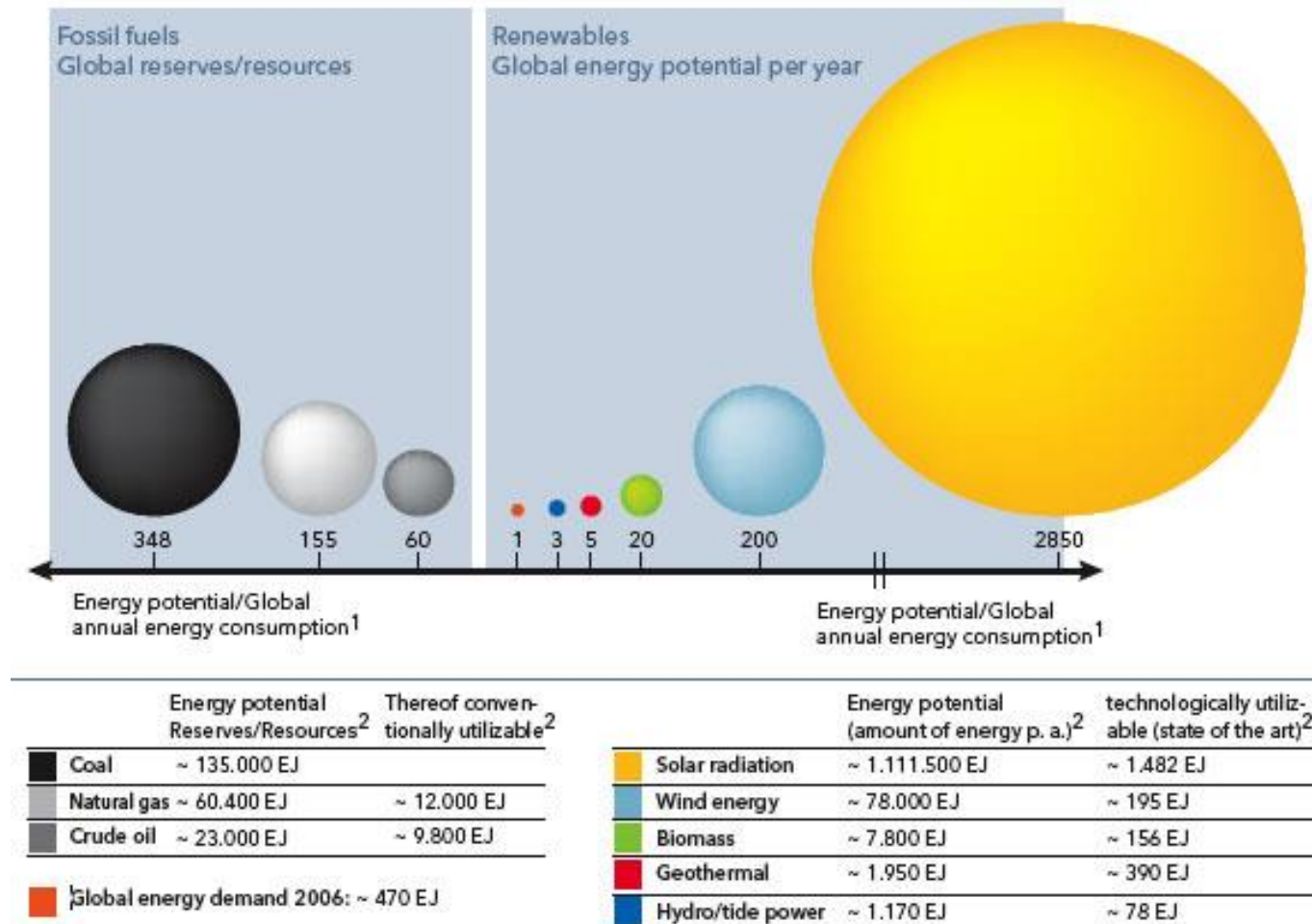
# Green valorization of biomass residues by solar-driven photoelectrochemical processes



Prof. Antonio Otavio de Toledo Patrocinio  
Laboratory of Photochemistry and Materials Science  
Institute of Chemistry – Federal University of  
Uberlândia, Brazil

[otaviopatrocínio@ufu.br](mailto:otaviopatrocínio@ufu.br)

# Solar Energy

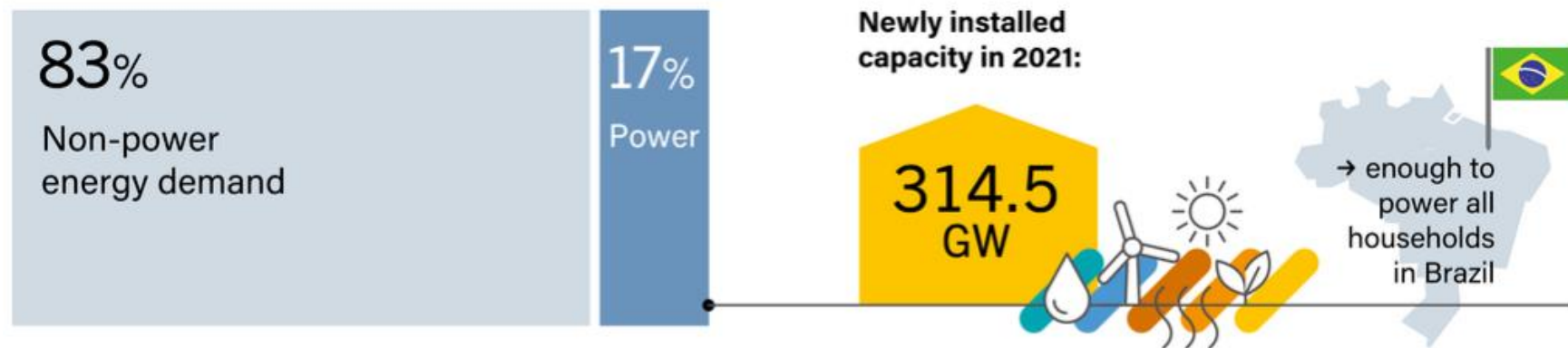


Data source: German Federal Institute for Geosciences and Natural Resources.

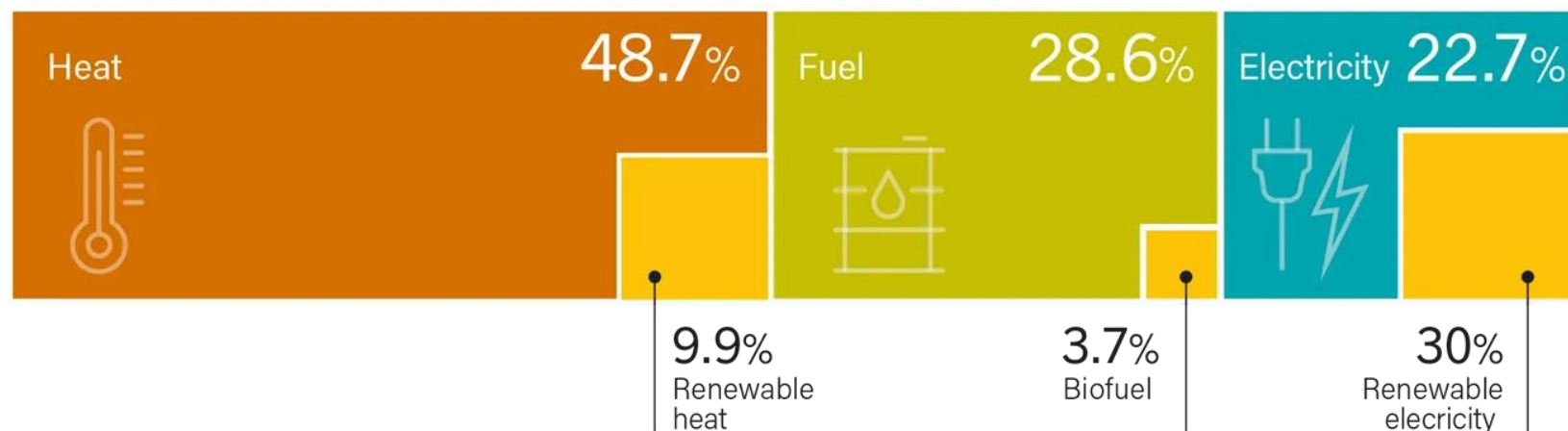
# Energy transition and solar fuels



Energy demand for power accounts for less than one-fifth of total final energy consumption

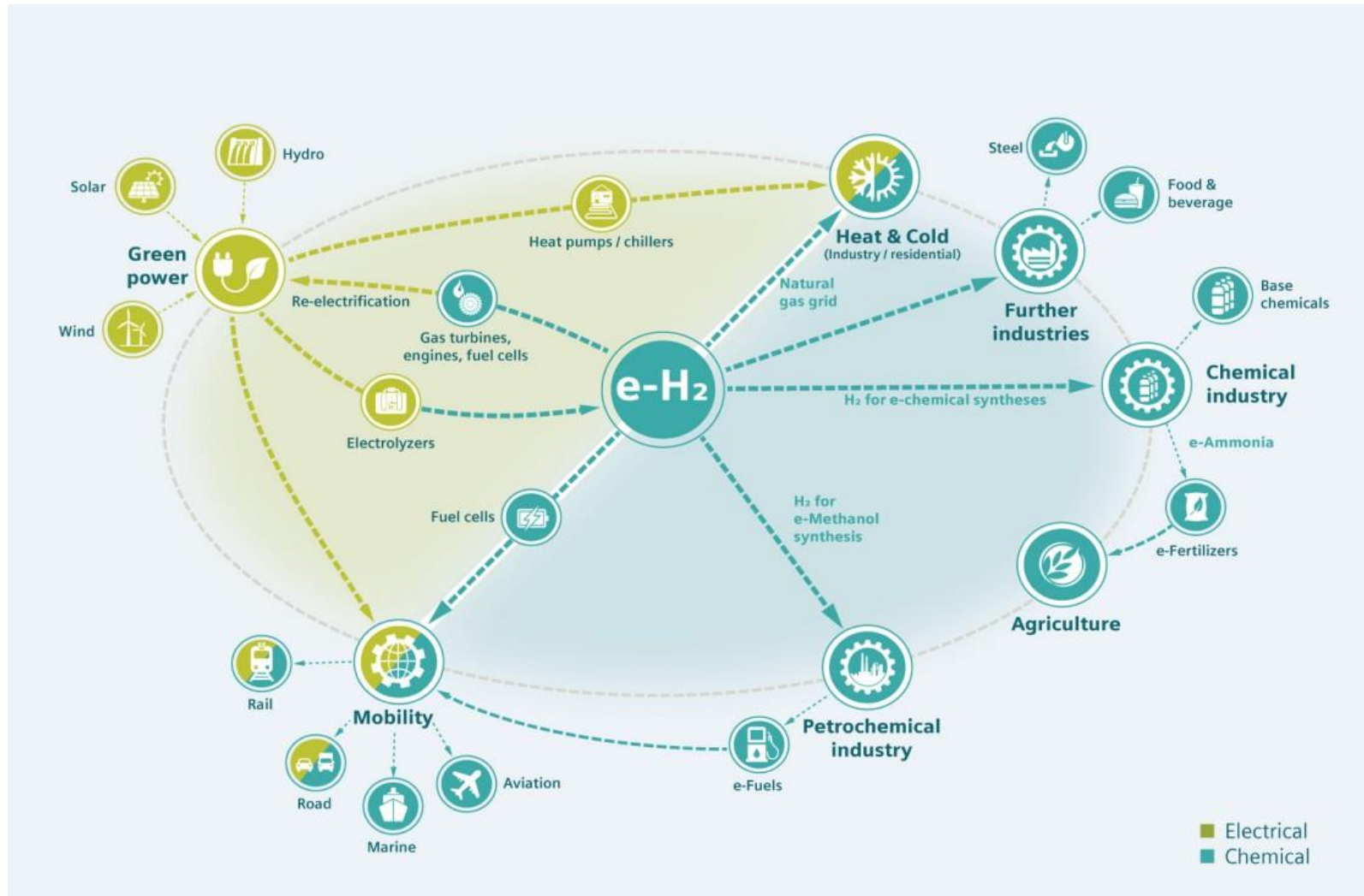


Total Final Energy and Total Modern Renewable Energy Share, by Energy Carrier, 2020

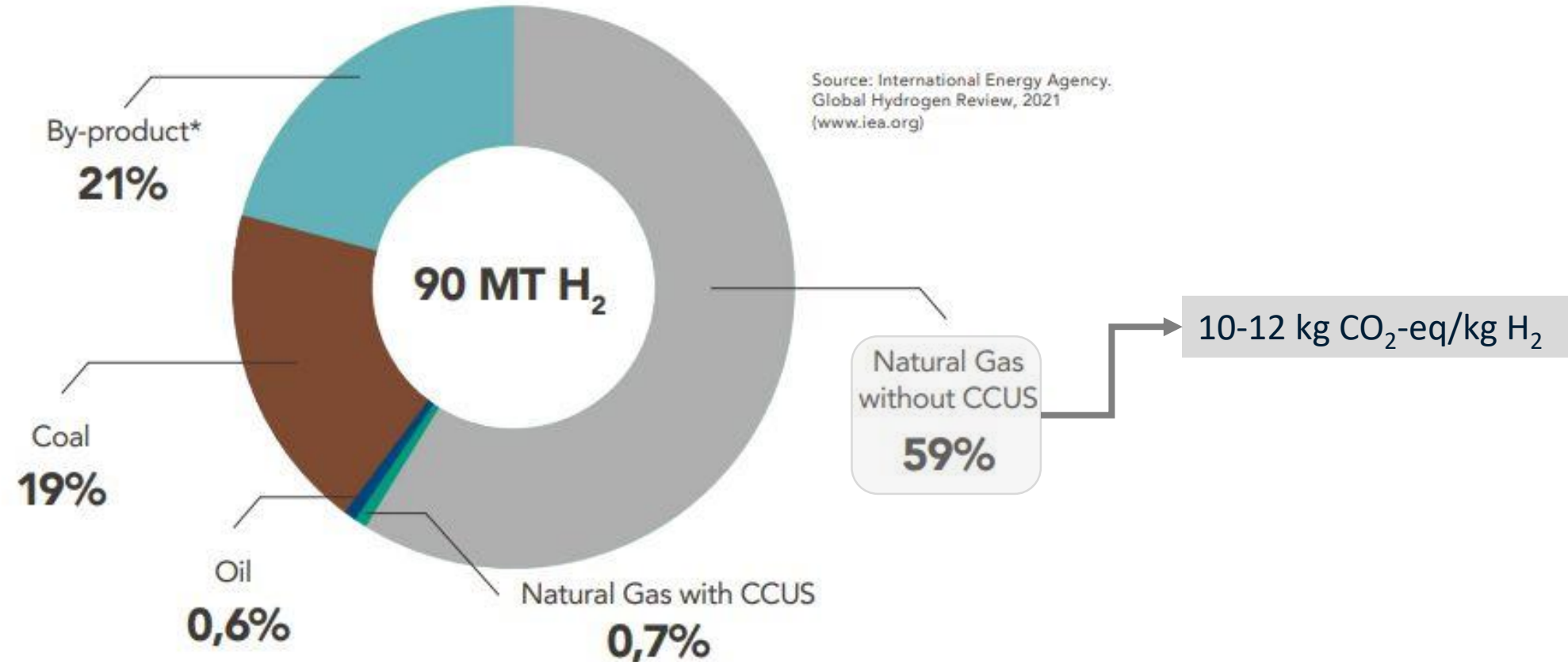


**Our main energy carriers are liquid or gaseous fuels**  
**We also need renewable feedstocks for the industries**

# H<sub>2</sub> as key energy vector



# Current H<sub>2</sub> industrial production methods



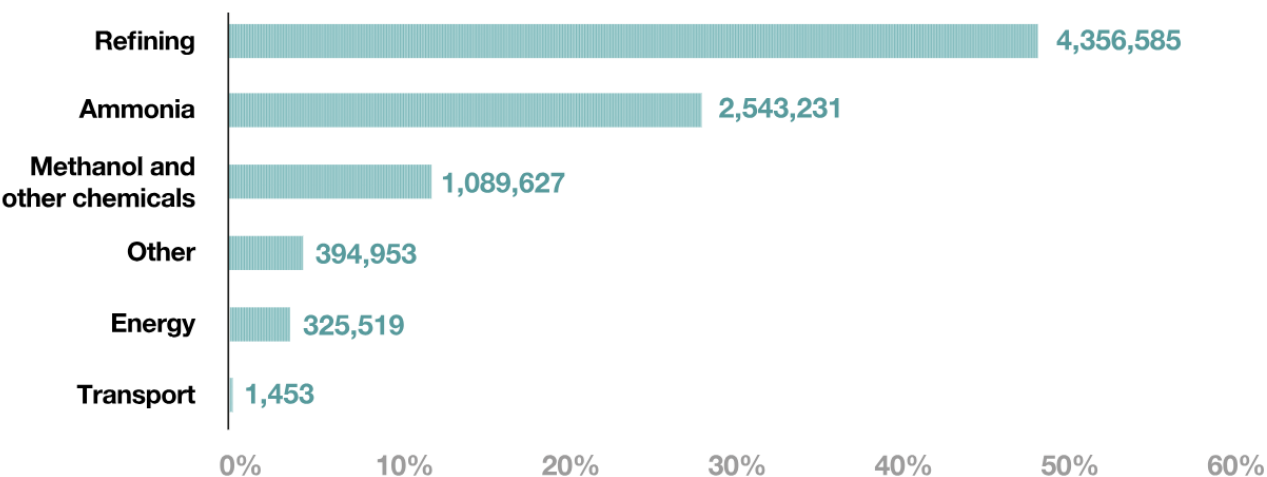
\*Hydrogen produced in facilities designed primarily for other products.



# Potential uses for low carbon hydrogen

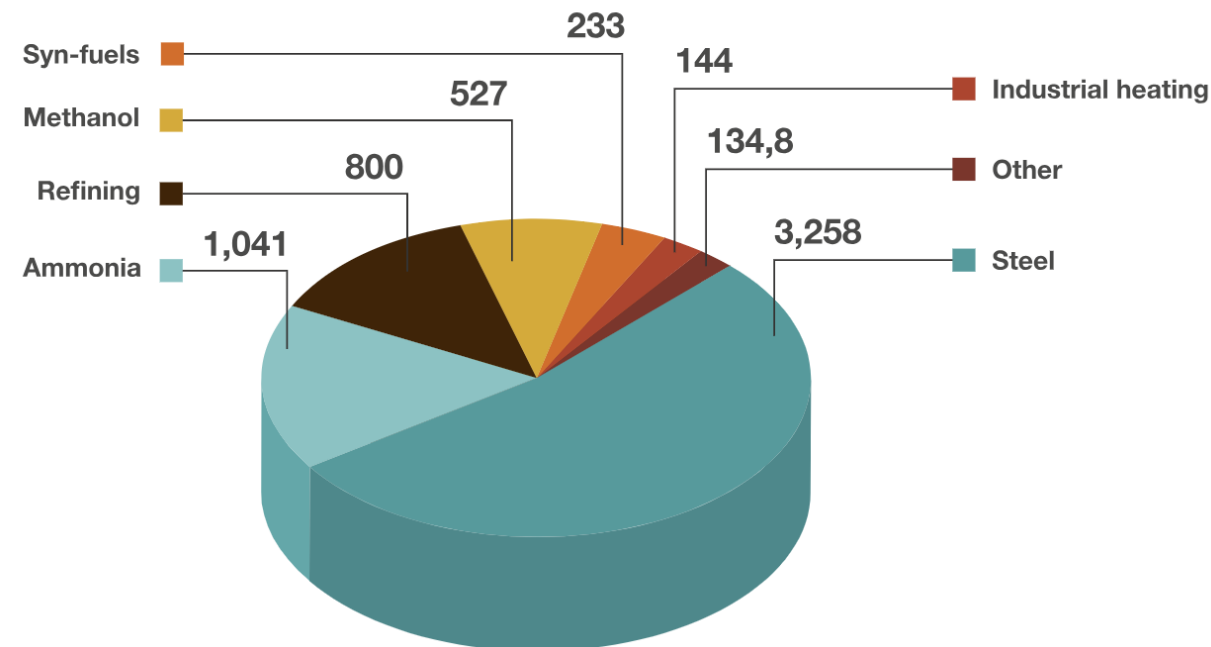


Total demand for hydrogen in 2020 by application



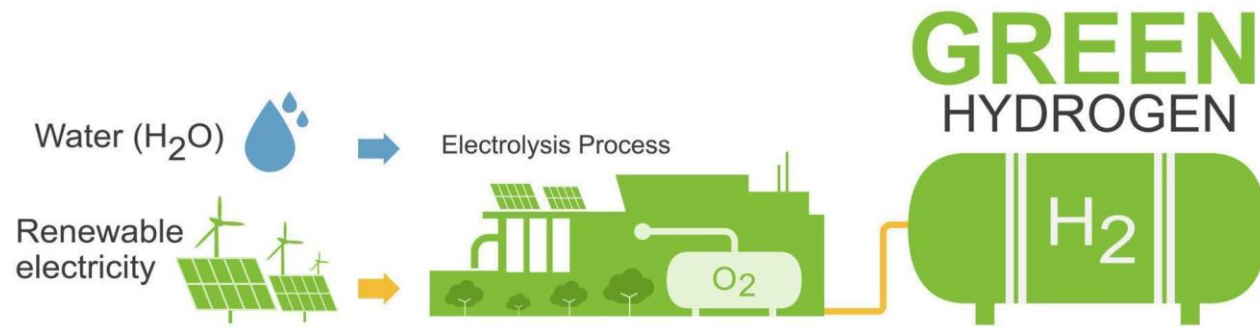
Source: Fuel Cells and Hydrogen Observatory

Planned clean hydrogen annual consumption in announced projects by 2030 by the industrial sector

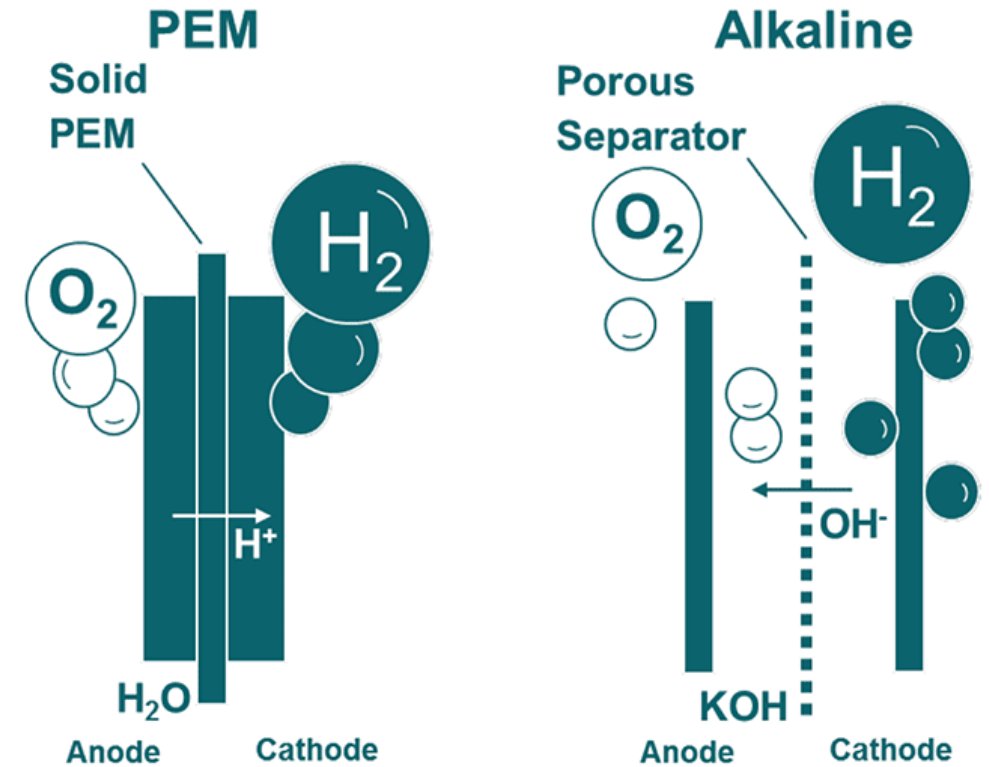


Source: Clean Hydrogen Monitor 2022, Hydrogen Europe

# How to produce low carbon hydrogen?



## Main electrolyzer technologies



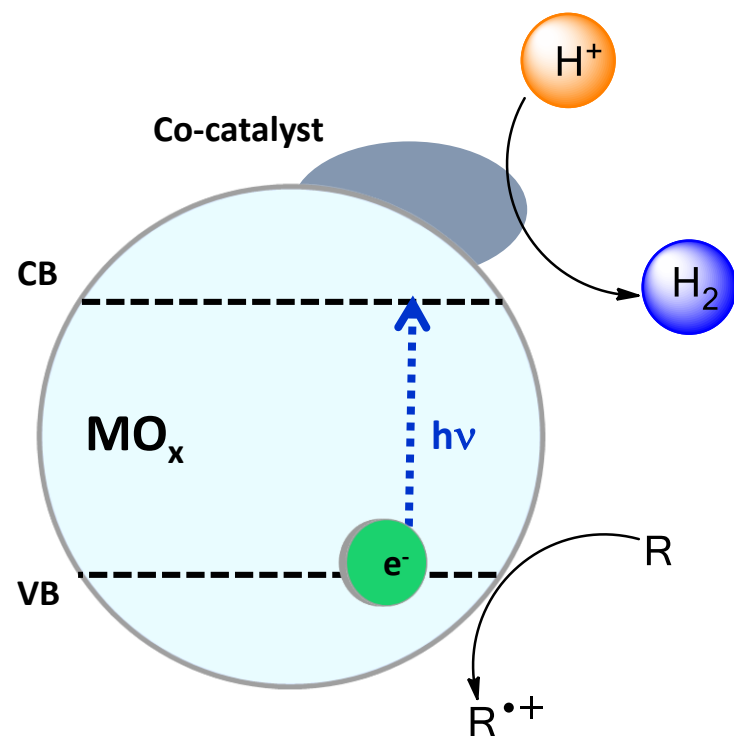
Mature technology but with higher production costs compared to natural gas reforming



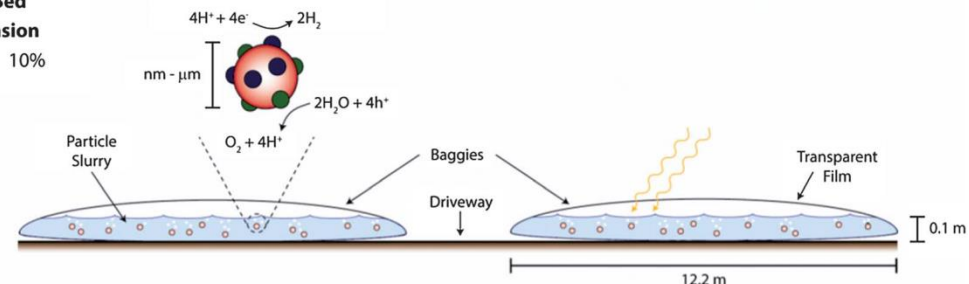
# Direct solar-to-hydrogen production



## Photocatalysis (PC)

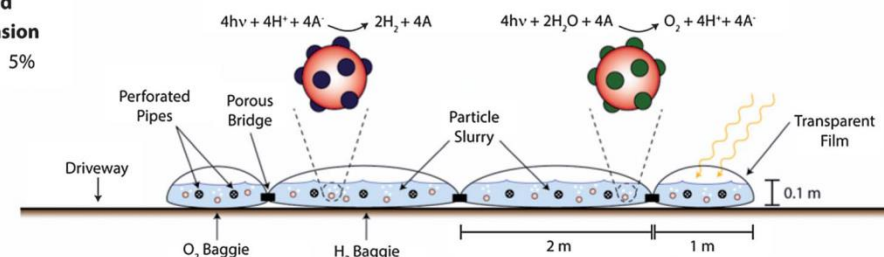


**Type 1: Single Bed Particle Suspension**  
STH Efficiency 10%



(a)

**Type 2: Dual Bed Particle Suspension**  
STH Efficiency 5%



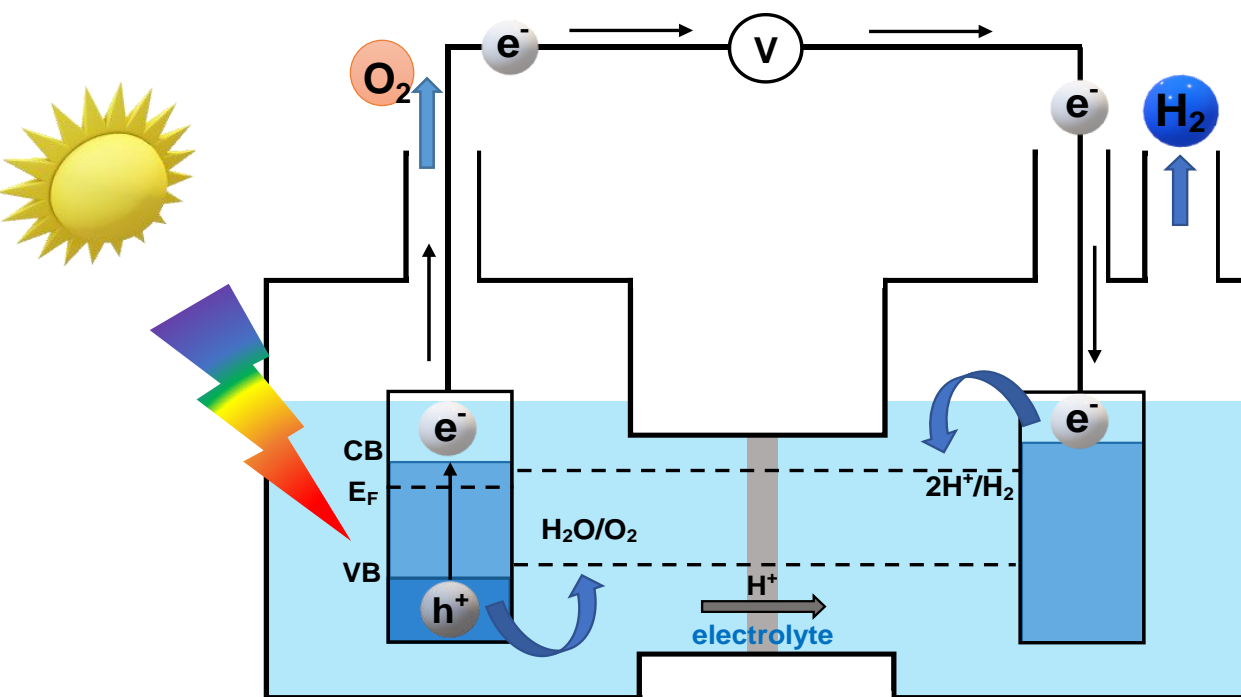
(b)



# Direct solar-to-hydrogen production

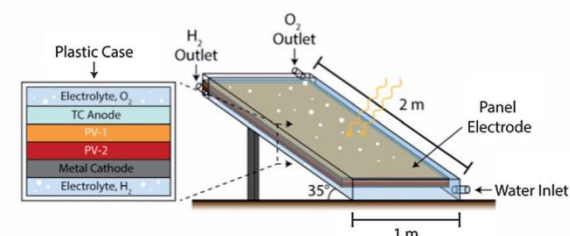


## Photoelectrochemical cells (PECs)



**Type 3: Fixed Panel Array**

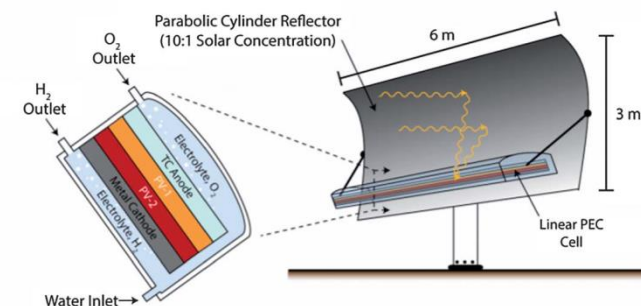
STH Efficiency 10%



(c)

**Type 4: Tracking Concentrator Array**

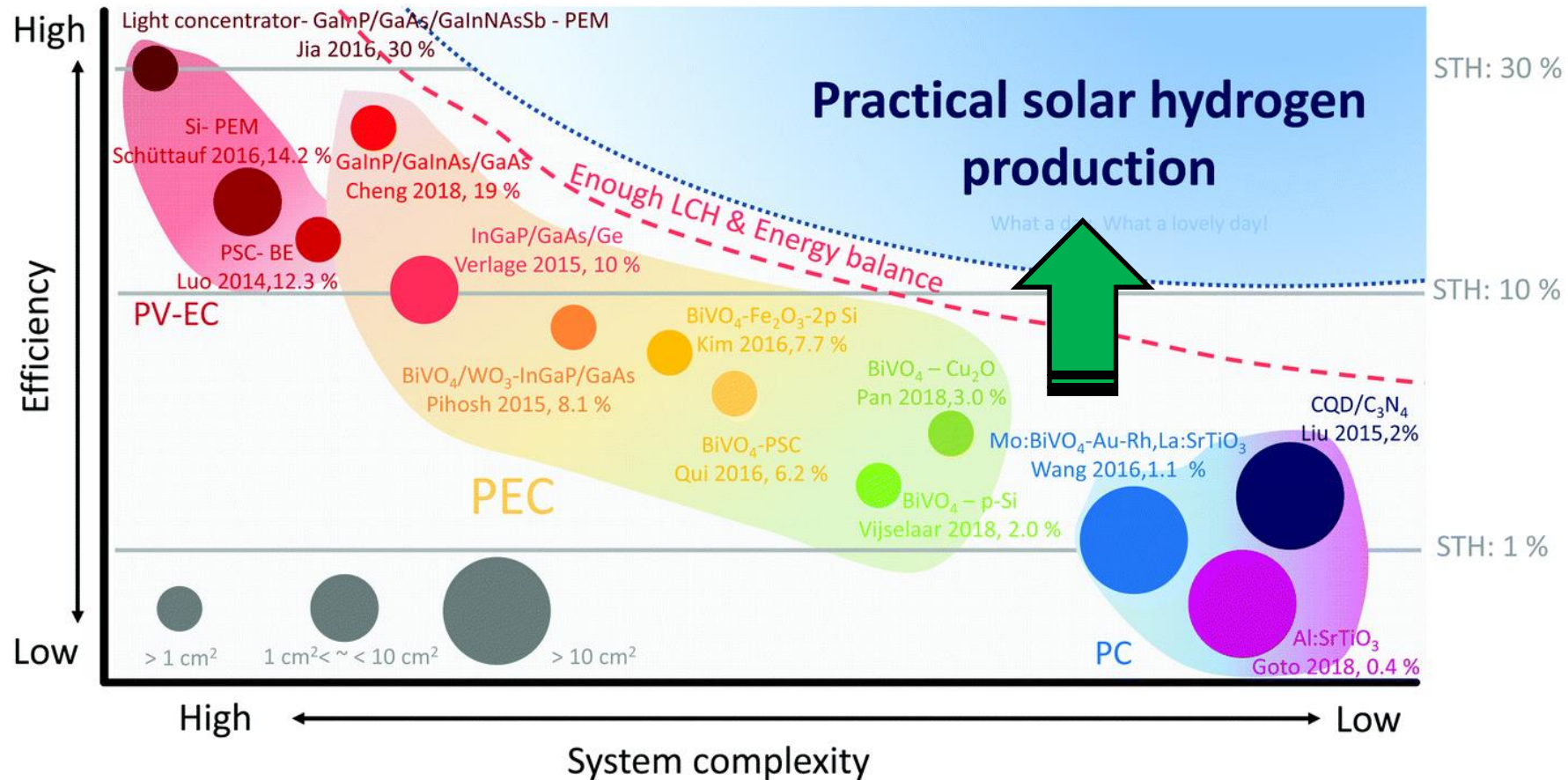
STH Efficiency 15%



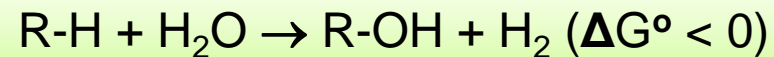
(d)

Energy Environ. Sci., 2013, 6, 1983–2002

# Challenges for upscaling



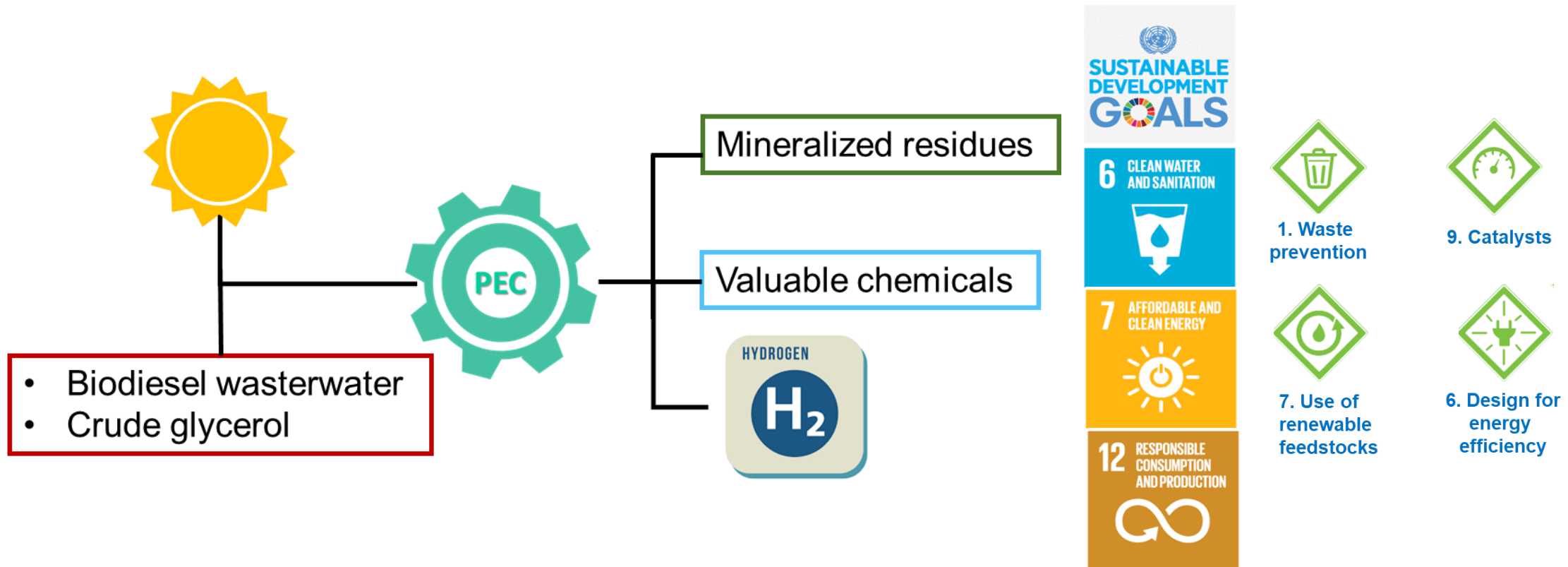
Water oxidation to molecular oxygen is the “bottleneck” for solar-to-hydrogen conversion



# Photoelectroreforming as an alternative for the water splitting

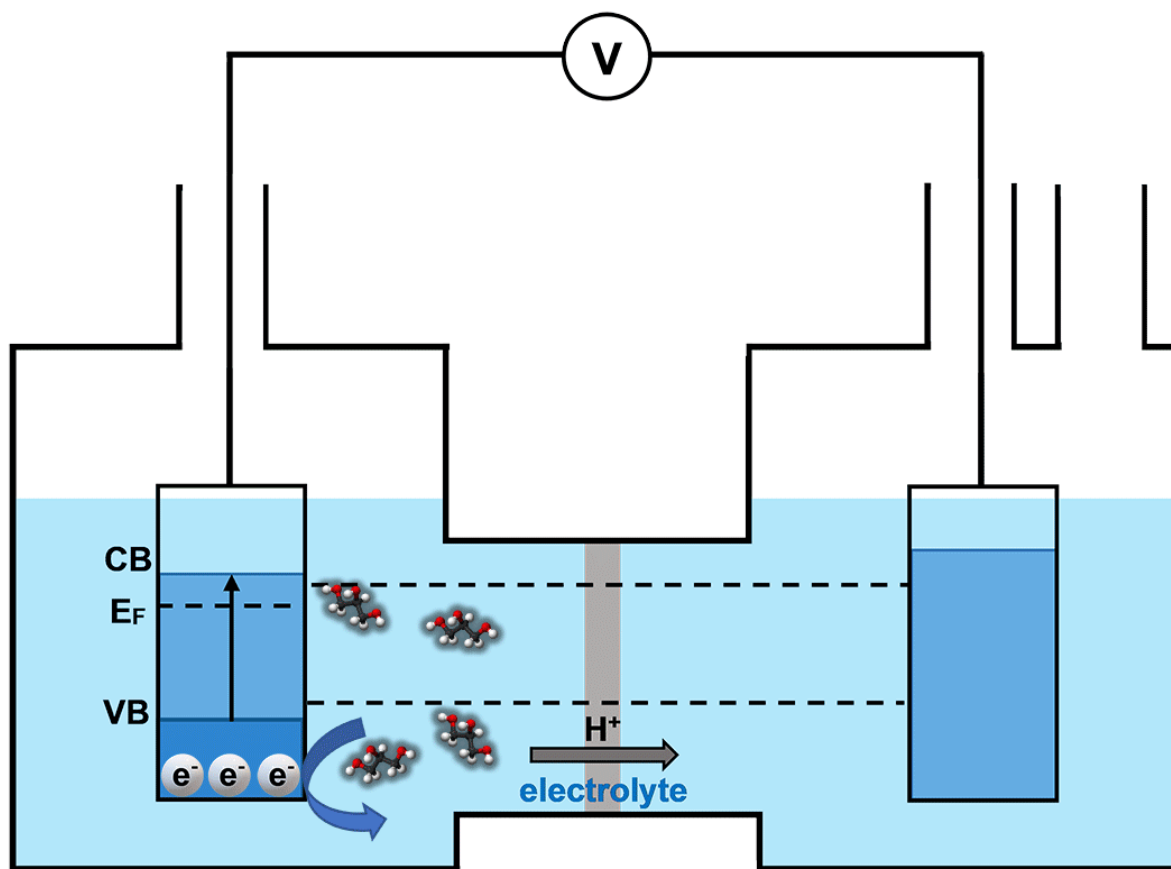


*Development of solar-driven photoelectrochemical cells for conversion of biomass residues into clean H<sub>2</sub> fuel and valuable chemicals*





# Biomass photoelectroreforming: how does it work?

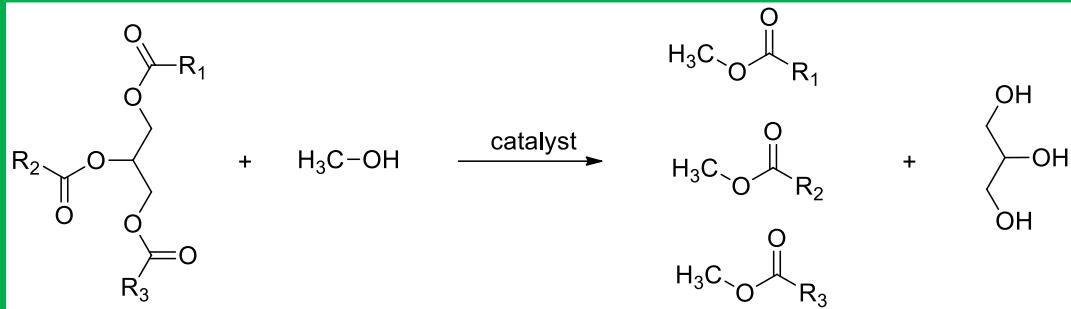




# Glycerol as substrate for photoelectroreforming?



*It is a byproduct of the transesterification of natural oil to produce biodiesel*



## Crude glycerol

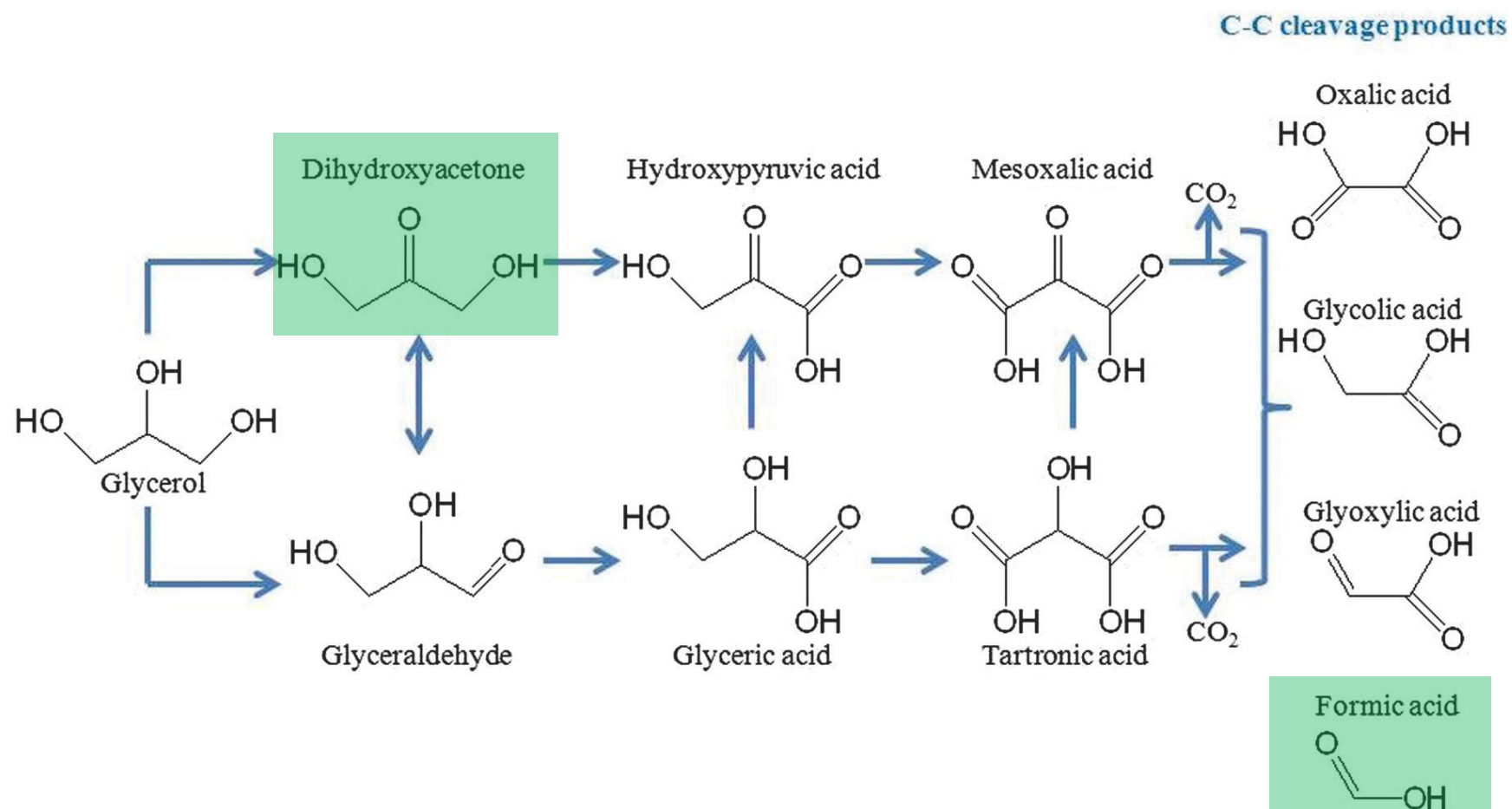
- ~10 m<sup>3</sup> of glycerol is produced for each 100 m<sup>3</sup> of biodiesel
- The production is 8 times higher than the current demand
- Purification is not attractive economically
- Storage, transportation and suitable destination is an economic and logistic issue



*Long term storage or incorrect disposal can lead to soil and natural water contamination*



# Glycerol as substrate for photoelectroreforming?



Glycerol is a suitable  
Chemical platform to  
obtain valuable  
organic molecules of  
industrial interest

# Glycerol photoelectroreforming – research strategy



## Green Chemistry for Life



### 2020 Green Chemistry for Life research grants

Dr Patrocínio Antonio Otávio de Toledo

Fundação de Apoio Universitário, Uberlândia, Brazil

for Project: "Green valorisation of biomass residues by solar-driven photoelectrochemical processes"

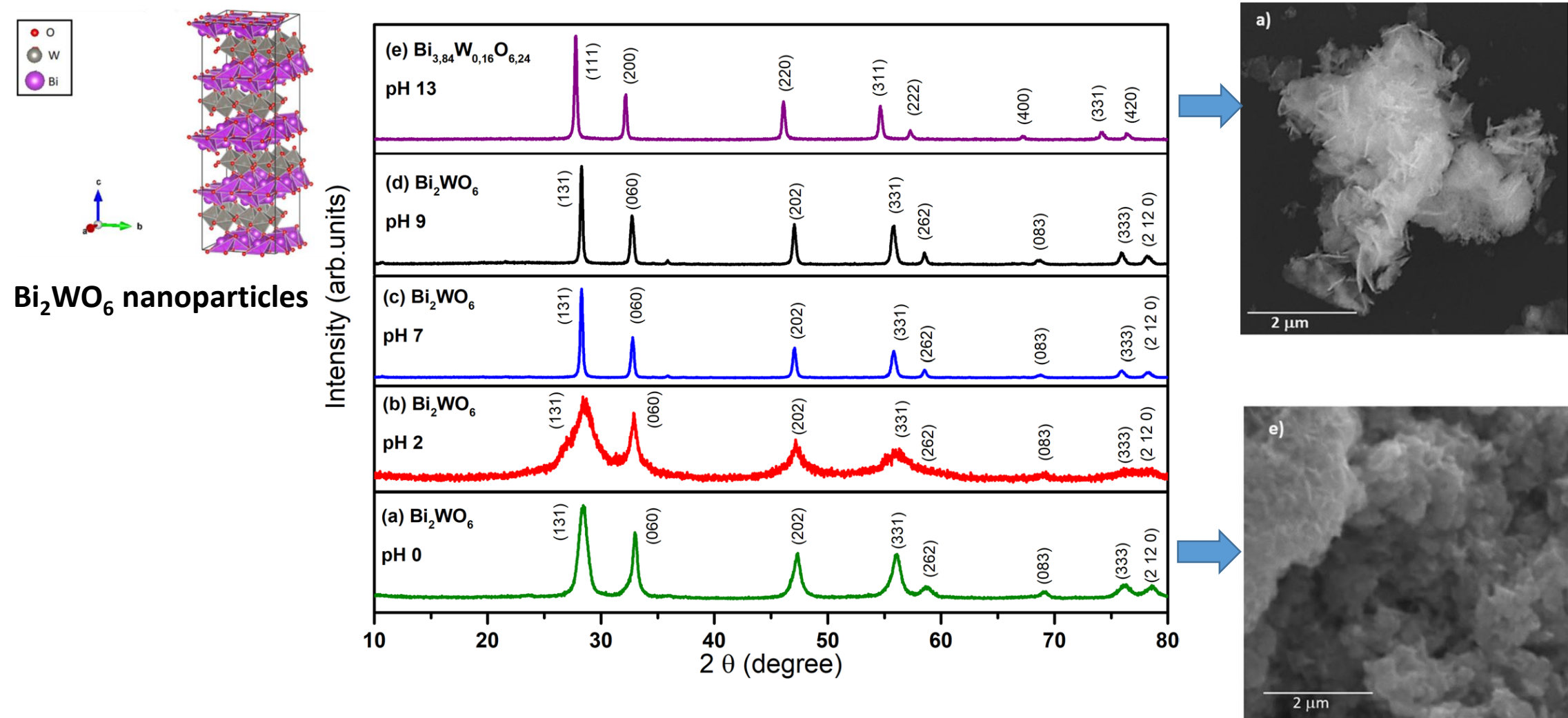
Synthesis and characterization of semiconducting nanomaterials based on Earth abundant elements

Development of cost-effective strategies for thin film deposition and device assembly

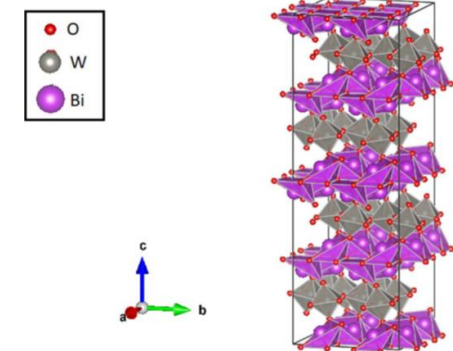
Optimization of the operation parameters aiming at high PEC efficiency and product selectivity

Long term stability tests and upscaling to pilot scale (TRL 7)

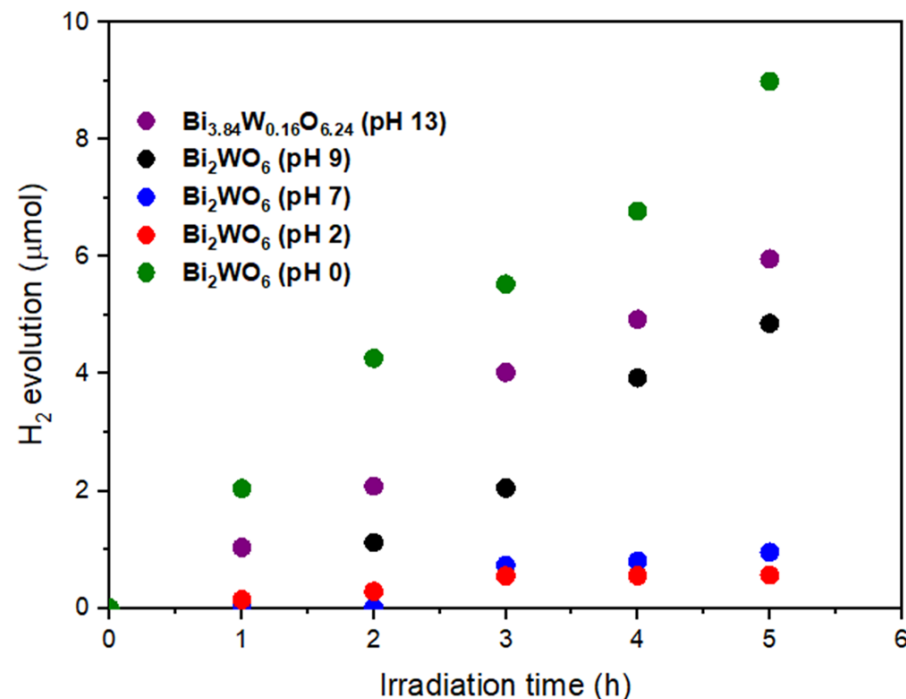
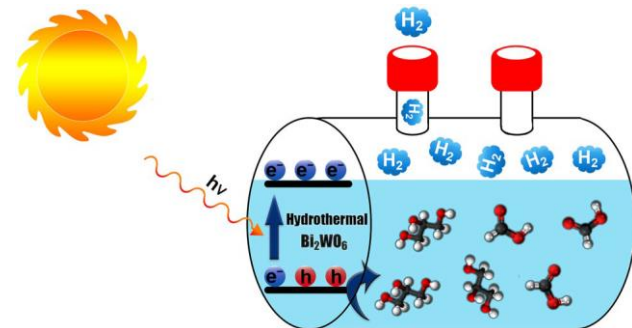
# Glycerol photoelectroreforming – some key results



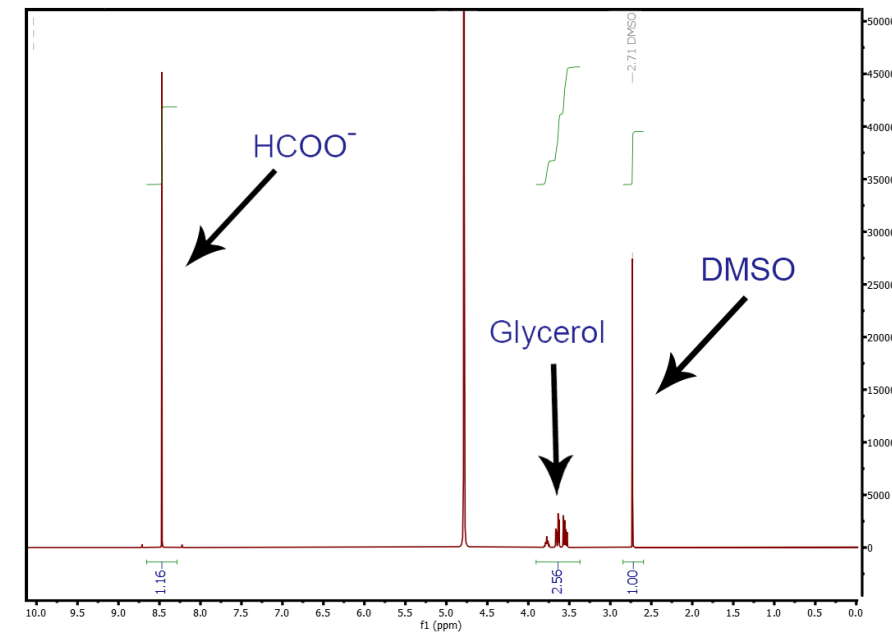
# Glycerol photoelectroreforming – some key results



$\text{Bi}_2\text{WO}_6$  nanoparticles



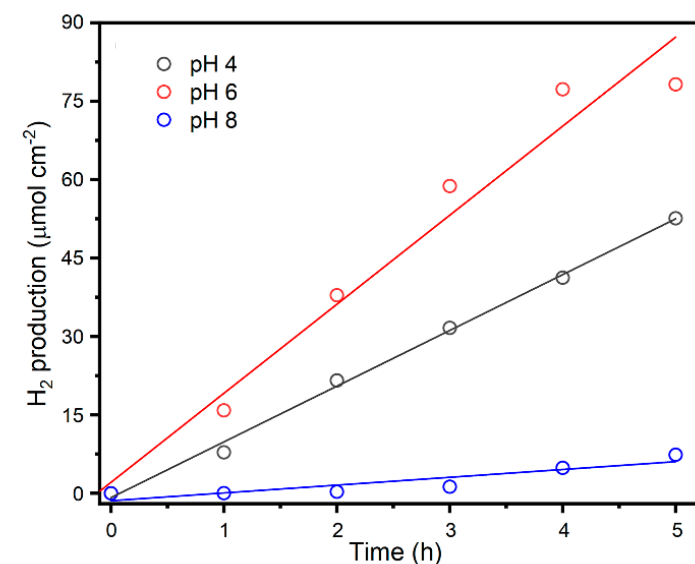
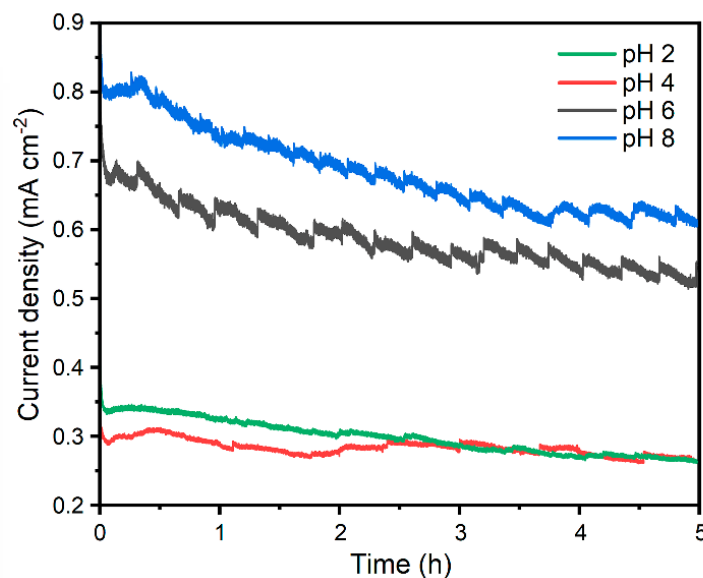
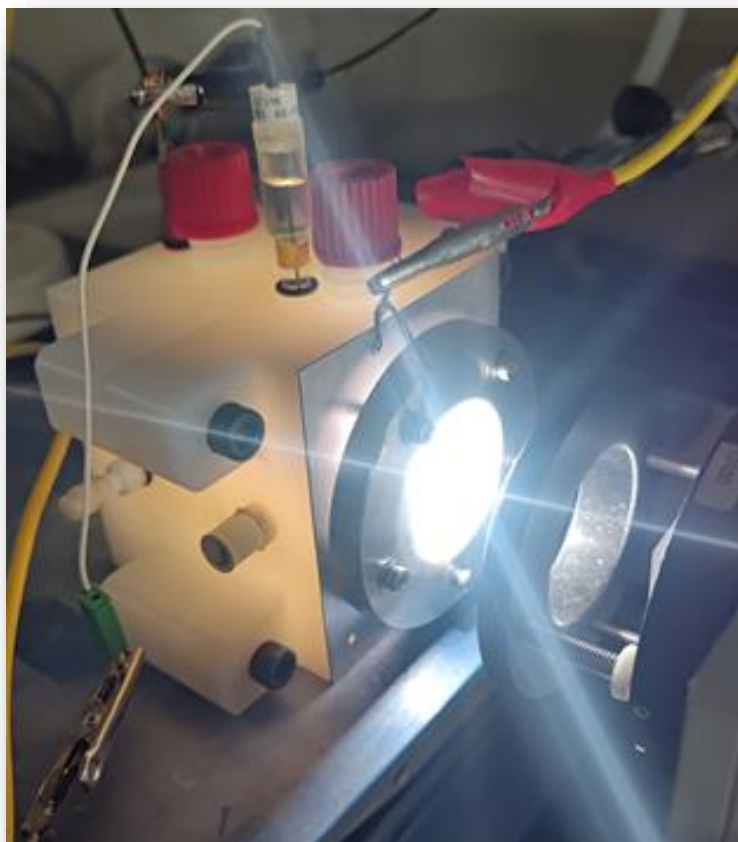
10 % Glycerol solution;  $\lambda > 350 \text{ nm}$  ( $100 \text{ mW cm}^{-2}$ )



Samples	Conversion	Product selectivity		$\xi \text{ H}_2$
		Formic acid	DHA	
$\text{Bi}_{3.84}\text{W}_{0.16}\text{O}_{6.24}$ (pH 13)	$63 \pm 6\%$	$87 \pm 2\%$	$<0.1\%$	$1.2 \pm 0.1\%$
$\text{Bi}_2\text{WO}_6$ (pH 9)	$49 \pm 5\%$	$41 \pm 2\%$	$0.1\%$	$1.0 \pm 0.1\%$
$\text{Bi}_2\text{WO}_6$ (pH 0)	$80 \pm 6\%$	$99 \pm 1\%$	$<0.1\%$	$1.4 \pm 0.1\%$

Photochemical & Photobiological Sciences 2022, 21, 1659

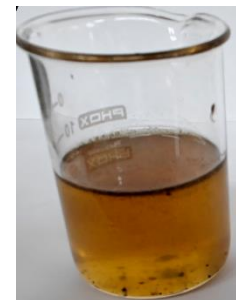
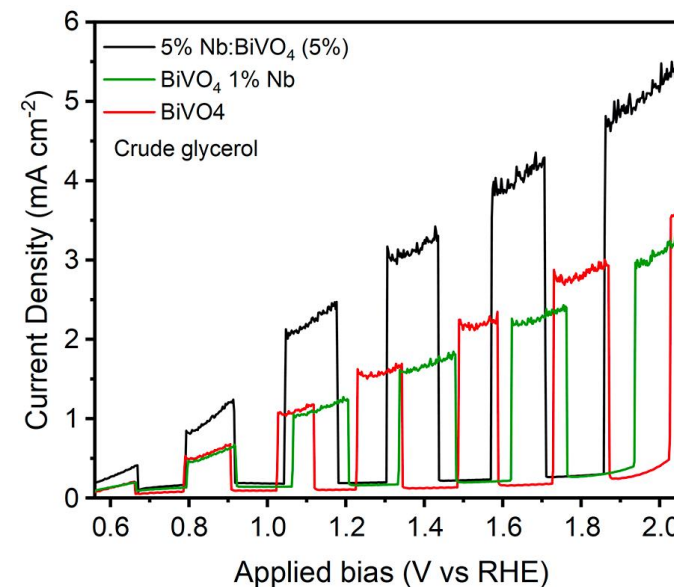
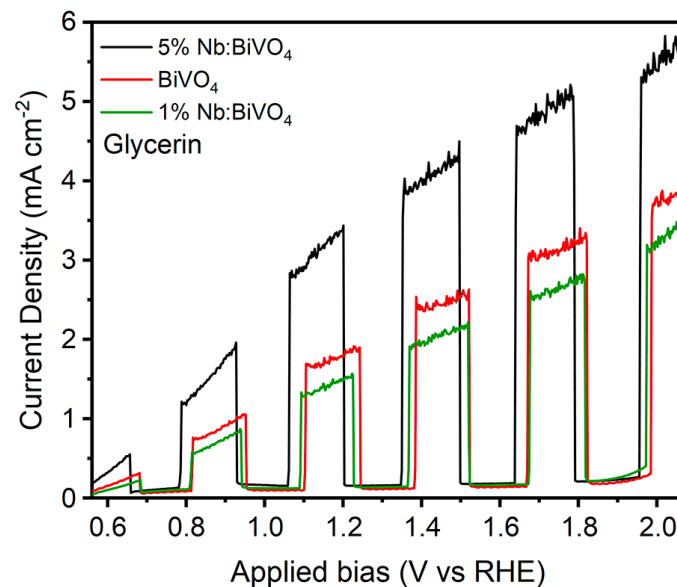
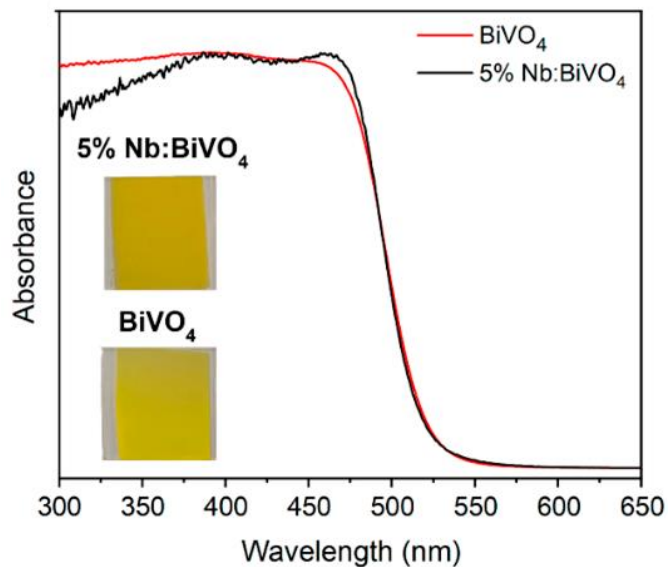
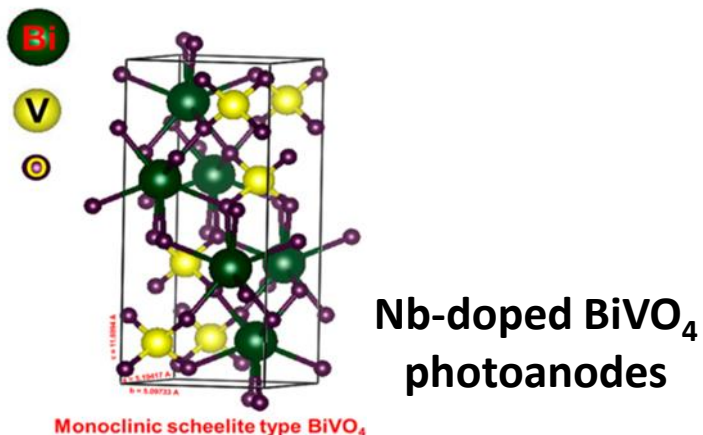
# Glycerol photoelectroreforming – some key results



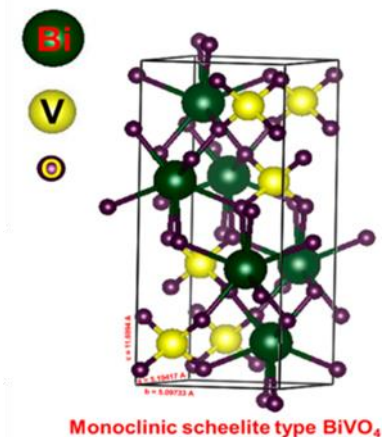
pH	Average current density (mA cm <sup>-2</sup> )	H <sub>2</sub> Production rate (μmol h <sup>-1</sup> cm <sup>-2</sup> )	F. E. (H <sub>2</sub> )	glycerol consumption	C-based selectivity (HCOOH)
4	0.29	10.7	95%	43%	88%
6	0.59	19.7	100%	65%	41%
8	0.70	2.0	17%	70%	13%



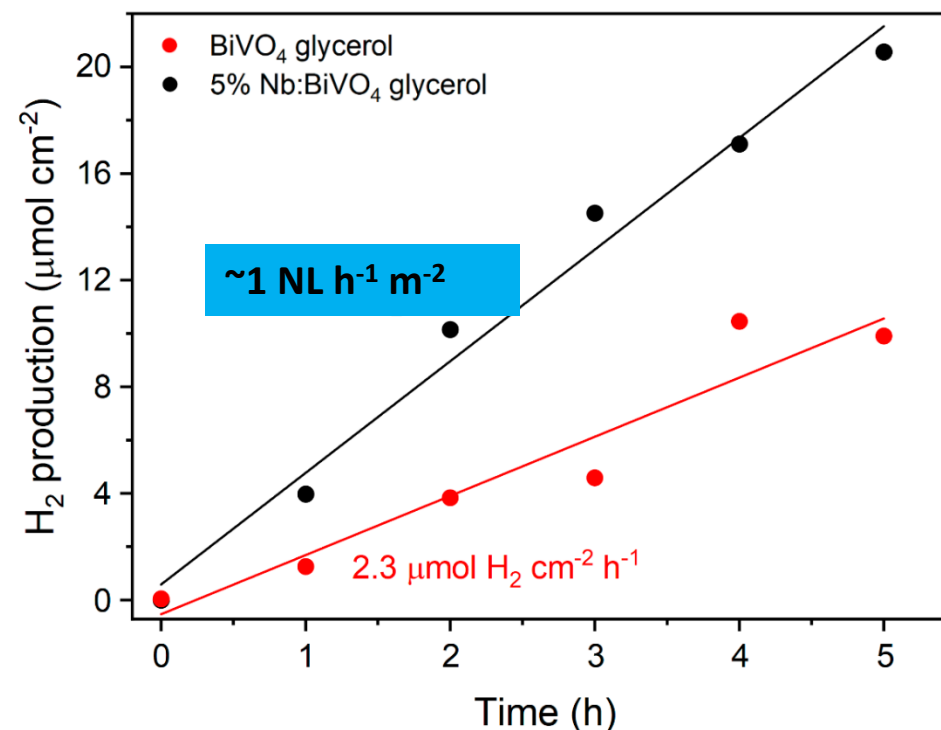
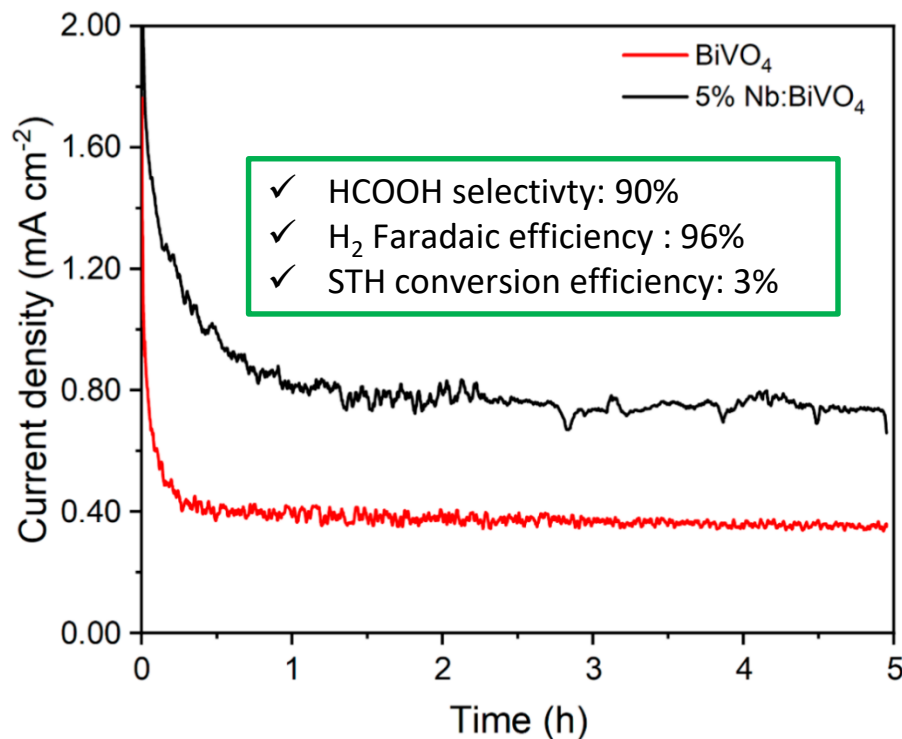
# Glycerol photoelectroreforming – some key results



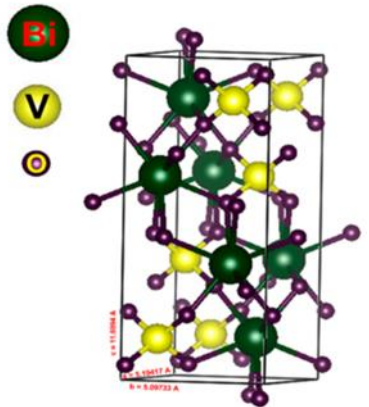
# Glycerol photoelectroreforming – some key results



Nb-doped  $\text{BiVO}_4$   
photoanodes

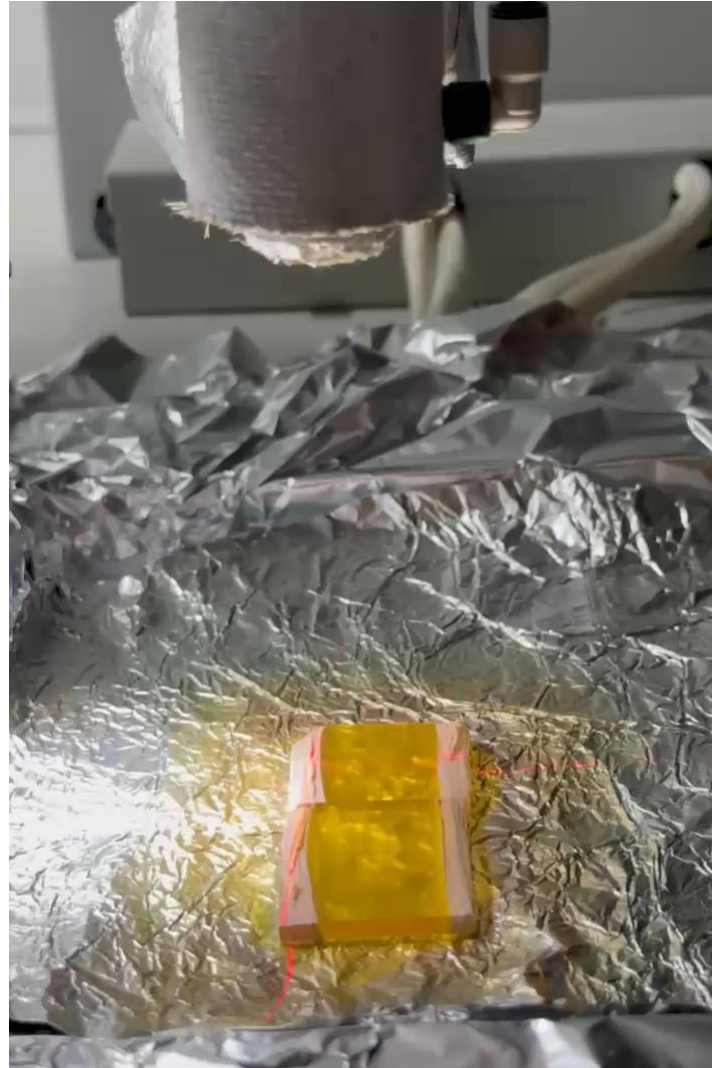
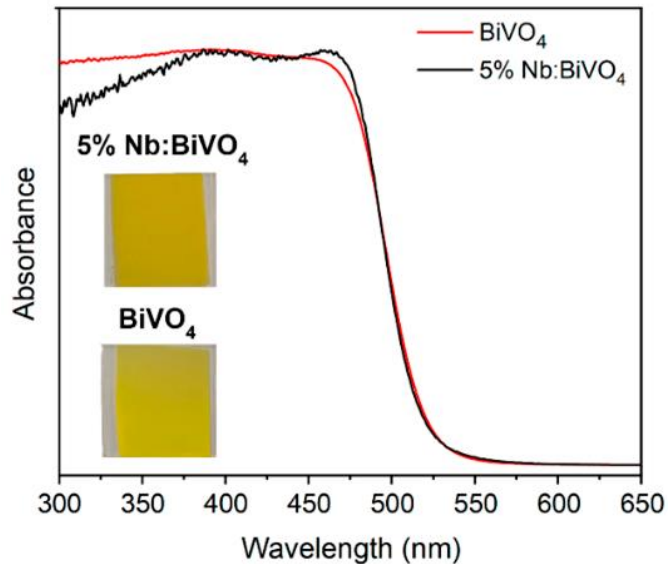


# Glycerol photoelectroreforming – some key results



Monoclinic scheelite type  $\text{BiVO}_4$

Nb-doped  $\text{BiVO}_4$   
photoanodes

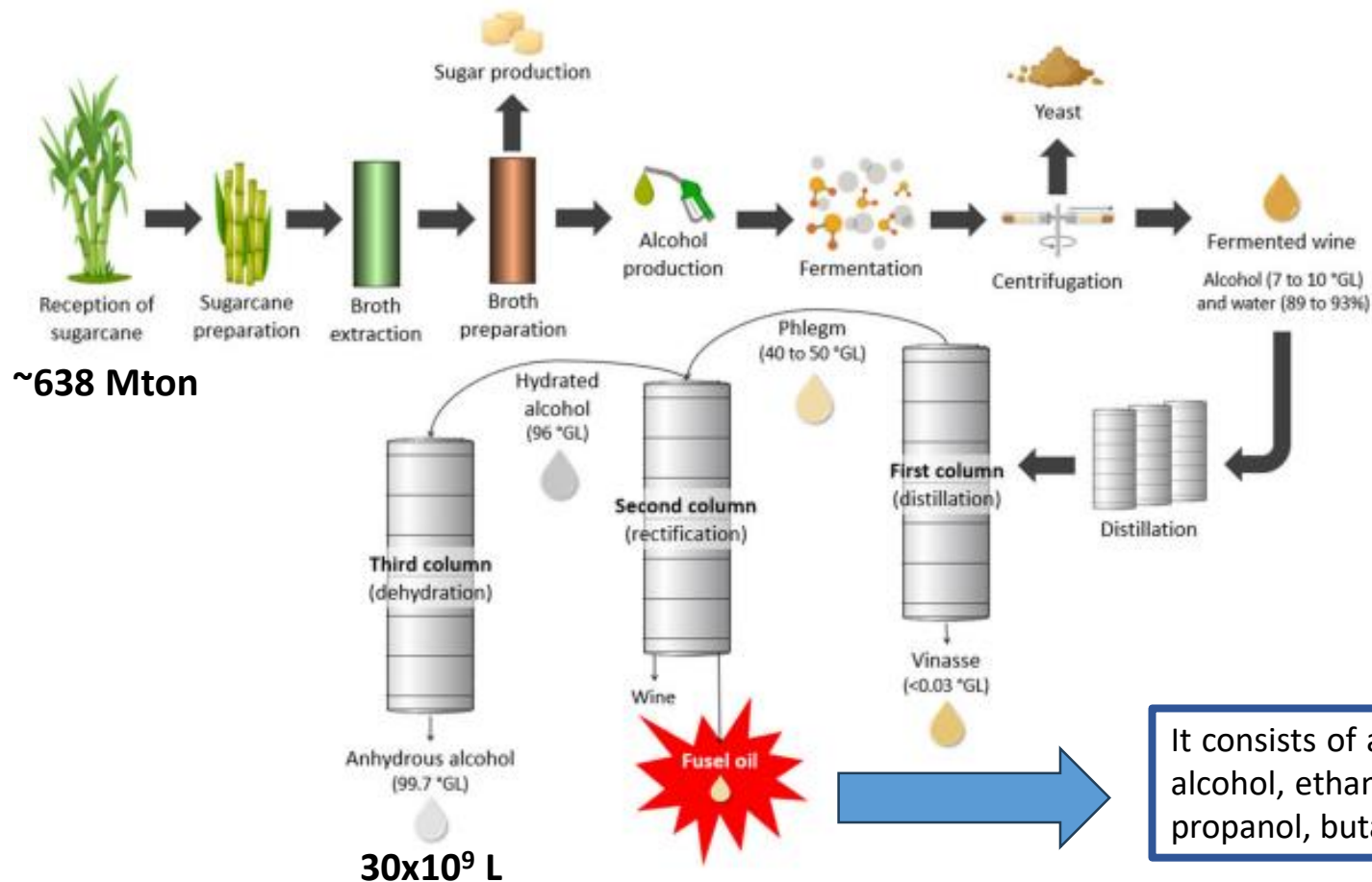


Homogeneous photoanodes through  
Computer-controlled deposition

# Can other residues be used for photoelectroforming?



## Bioethanol production in Brazil

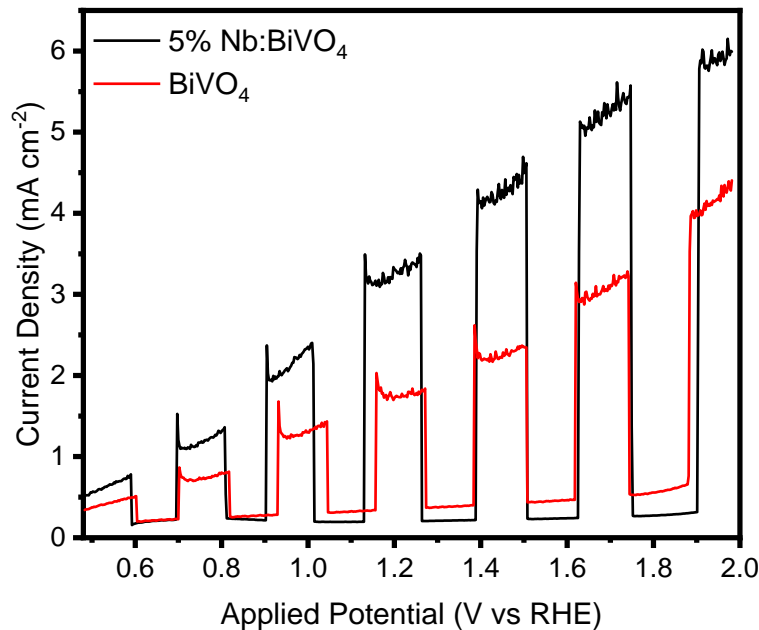


For 100 L of ethanol, it is also produced:

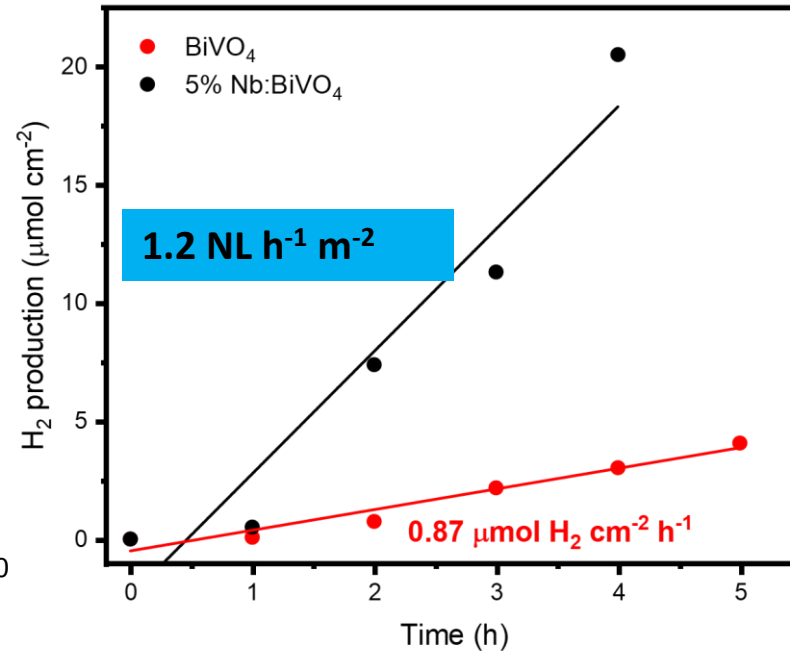
- 1000 L of vinasse
- 0,2 L of fuseloil
- 3 L of flegmas (diluted fuseoil)

It consists of an azeotropic mixture between C3-C5 alcohols, mainly isoamyl alcohol, ethanol and water and others trace compounds such as isobutanol, propanol, butanol, furfural.

# Flegmass photoelectroreforming



**Flegmass 100%**, 0.1 mol/L K<sub>2</sub>SO<sub>4</sub>  
1000 W m<sup>-2</sup> A.M. 1.5G



- ✓ Optimal performance at neutral pH
- ✓ High selectivity to formic acid (90%)
- ✓ Performance 5 times better than TiO<sub>2</sub> (Standard system)
- ✓ Long term stability thanks to the Nb(V) doping
- ✓ Reduced water use



# Conclusions and perspectives

---



- Low carbon H<sub>2</sub> production can play a key role on the decarbonization of industrial processes
- Its production costs need to be reduced to efficiently compete with the current production methods (natural gas reforming / coal gasification)
- Direct solar-to-hydrogen conversion can be a suitable method to reduce the costs of low carbon H<sub>2</sub>
- The use of biomass residues as substrates for H<sub>2</sub> production is an energy efficient pathway that can also lead to the production of valuable organic substrates increasing the economic attractiveness of the process.
- Further R&D is needed to scale-up the photoelectrochemical cells

# Acknowledgments



INTERNATIONAL UNION OF  
PURE AND APPLIED CHEMISTRY





---

**Thank you!**